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STUDY. PLAY. Kinetic Theory of Gases. 1.

Gases are made up of molecules with lots of empty space between. 2. Particles move quickly - pressure is caused by this motion 3. Collisions are elastic 4.

There are no attractive or repulsive forces between molecules 5.  $KE = \frac{1}{2} M \cdot V^2$

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Chemistry Chapter 13. carbohydrates (saccharides) metabolic fuel (function of carbs) monosaccharide (s) (simple sugars) disaccharide (s) polyhydroxy aldehydes or ketones or molecules that yield compo.... simple sugars serve as an immediate source of energy while com....

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Review 13-1 Representing Sample

Spaces DATE BLOCK Represent a

Sample Space The sample space of an experiment is the set of all possible

outcomes A

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CHAPTER 13 Table Of Contents Click a hyperlink to view the corresponding slides. • State the relationships among pressure, temperature, and volume of a constant amount of gas.

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